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Algebra 2 P Chapter Practice Test Name: Please show your work. You are allowed only a basic calculator. This is last years test.1. Perform the indicated operations and simplify 2a)  $(5x^2 - 7y)$  b)  $(2x + 7)(3x^2 - 8x + 1)$ 2. Factor each expression completely a)  $x^2 - 3x - 40$  b)  $2x^3 + x^2 - 6x - 3$  c)  $x^4 + 5x^3 + 6x^2$  d)  $2(a + b)^2 + 5(a + b) - 33$ . a) Write  $x < 8$  in interval notation and...

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### [UNIT 4 PRACTICE TEST](#)

Honour Chemistry: Unit 4 Practice Test: Thermochemistry and Nuclear ChemistryPart A: Multiple Choice and Numerical Response (1 mark each for Multiple Choice) (2 marks each for Numerical Response)  $E_{\text{bind}} = \sum mc^2$   $c = \text{speed of light}$  (3.00 10<sup>8</sup> m s<sup>-1</sup>) 1 eV = 1.69 10<sup>-19</sup> J (1 kg = 6.022 10<sup>26</sup> amu , or 1 g = 6.022 10<sup>23</sup> amu = 1 mole amu) 1 MeV = 1.69 10<sup>-16</sup> J (1 Neutron = 1.008665 amu) (1 Proton = 1.007825 amu)

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### [UNIT 2 \(PART 1\) PRACTICE TEST](#)

Honour Chemistry Practice Test: Unit 2 (Part 1): Matter as Solutions and Gases 1 atm = 760 mm Hg = 760 torr = 101.325 kPa  $T_K = T_C + 273.15$  STP = 22.4 L mol @ 0.00C & 101.325 kPa (1 atm) SATP = 24.8 L mol @ 25.0C & 100. kPa L atm L kPa m m nsolute  $R = 0.0821$  or  $8.314$   $D = n = C = K$  mol K mol V M Vsolution  $P_1V_1 = P_2V_2$   $r_1M_2 = n_1P_1 = C_1V_1 = C_2V_2$   $PV = nRT = \sum P_iV_i = \sum n_iRT_i$   $r_2M_1 = n_{\text{Total}}PT_{\text{Total}}$   $PT_{\text{Total}} = P_1 + \dots$

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### [PRACTICE TEST 7](#)

Practice Test 7, solutions belowProblem 11. (p & q) A2. (p v q) A3. Show: rProblem 21. (p ? q) ? (r ? s) A2. p v q A3 (q & p) A4. Show: (r v s)Problem 31. (p v q) v ( r v s) A2. r A3. (t ? s) A4. Show p ?q Problem 11. (p & q) A2. (p v q) A3. Show: r4. p & q &, 25. p & q , 46. p & q , 57. r !, 1,6Problem 21. (p ? q) ? (r ? s) A2. p v q A3 (q & p) A4. Show: (r v s)5. (r v s) A1P6. p ? q ?v, 27. q v ...

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